

1D MULTIPHASE MODELING OF NO_x AND HALOGEN PHOTOCHEMISTRY AT SUMMIT, GREENLAND

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AND

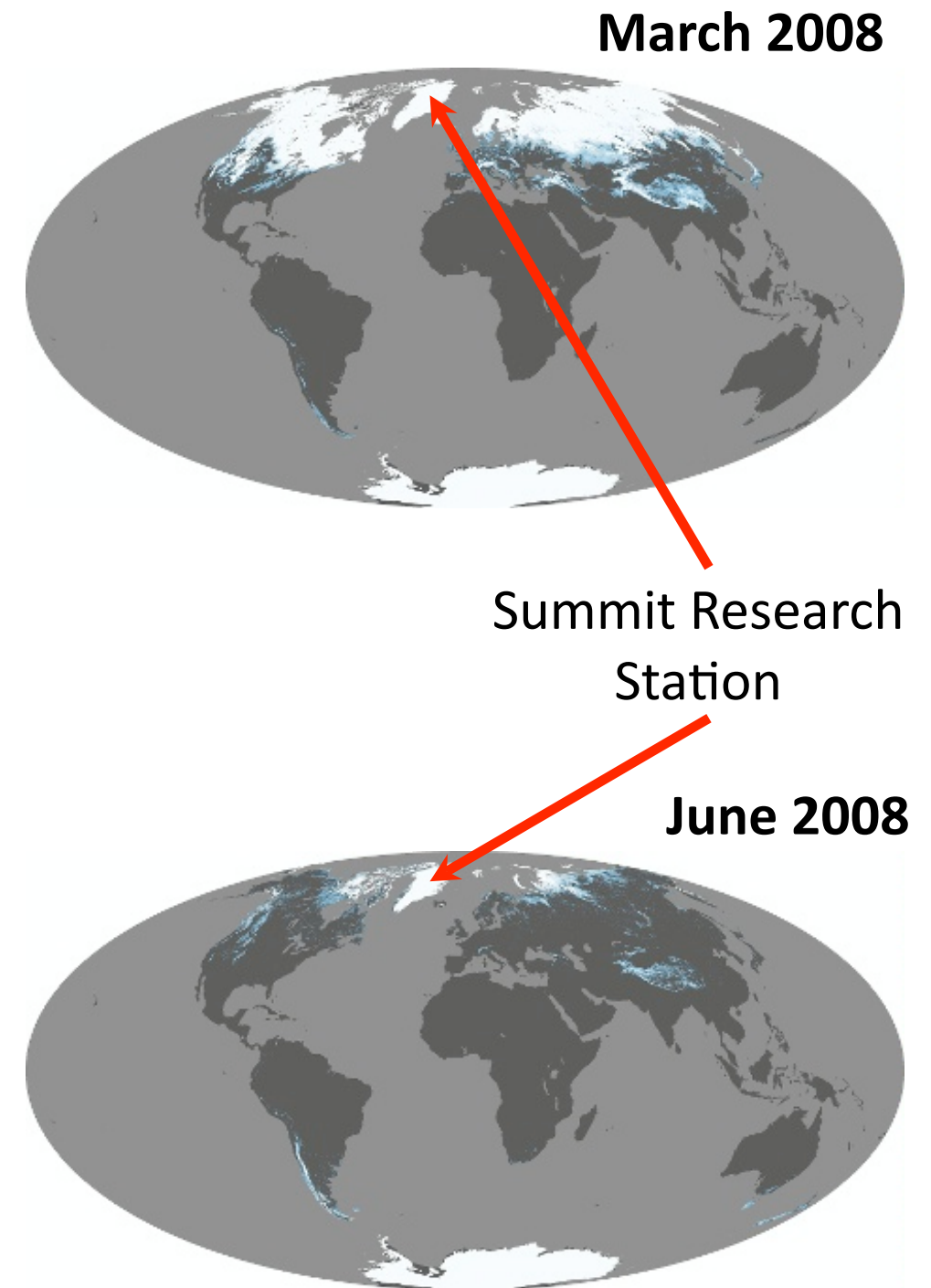
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AIR-ICE CHEMICAL INTERACTIONS WORKSHOP

JUNE 6, 2011

Why study air and snow photochemistry in remote regions?

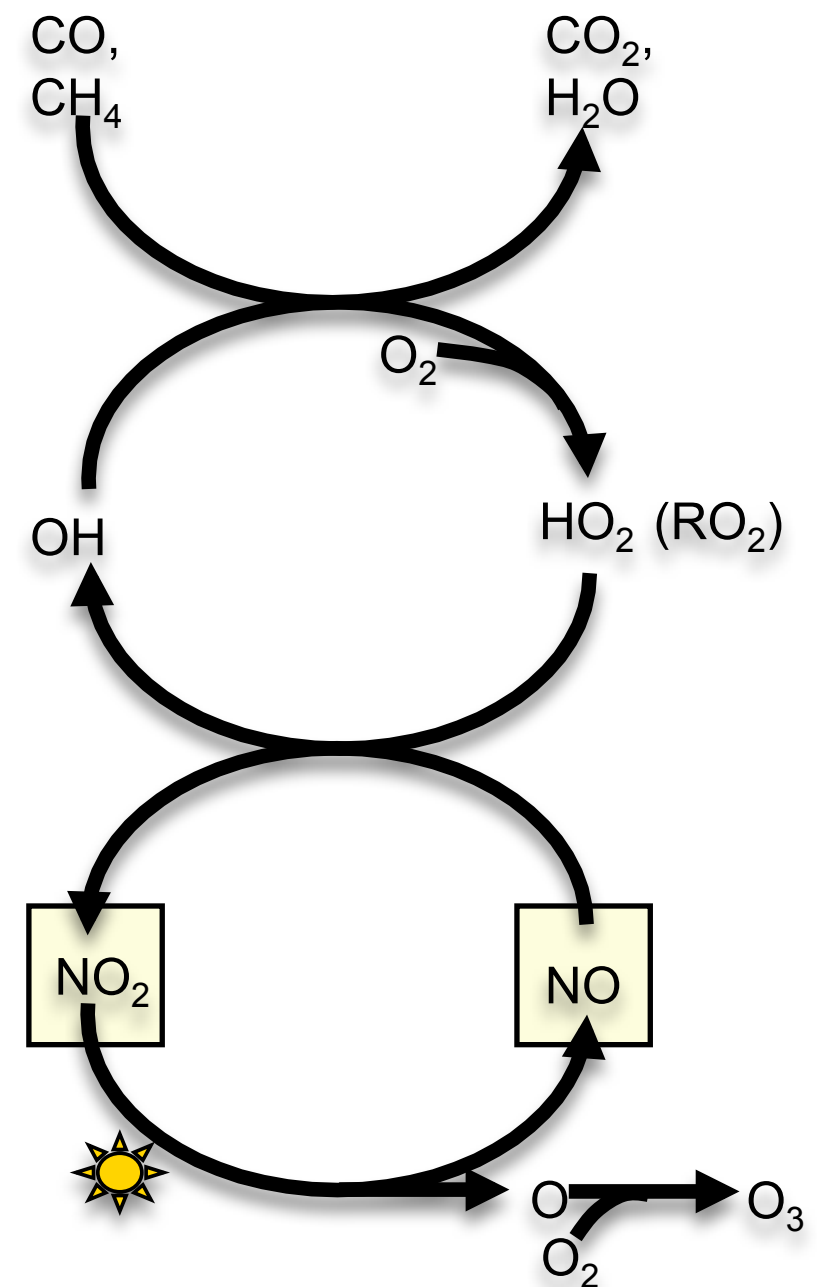
- Recent work shows that photochemistry in and above snow can play an important role in Arctic and Antarctic atmospheric chemistry
 - **Ozone** destruction in the Arctic MBL and in interstitial air
 - **NO_x** emission from snow
 - **Hg** deposition
 - Variation in **oxidative capacity**, i.e. OH/HO₂ levels
- **Questions:**
 - Does snowpack chemistry impact oxidative capacity, ozone, Hg, methane and HC lifetimes, etc.?
 - What role do snow surfaces play? Are all snow surfaces photochemically active?
 - **Does snow photochemistry impact atmospheric composition on larger scales?**



Snow cover from NASA Earth Observations (NEO) using data from the MODIS sensor on NASA's Terra satellite

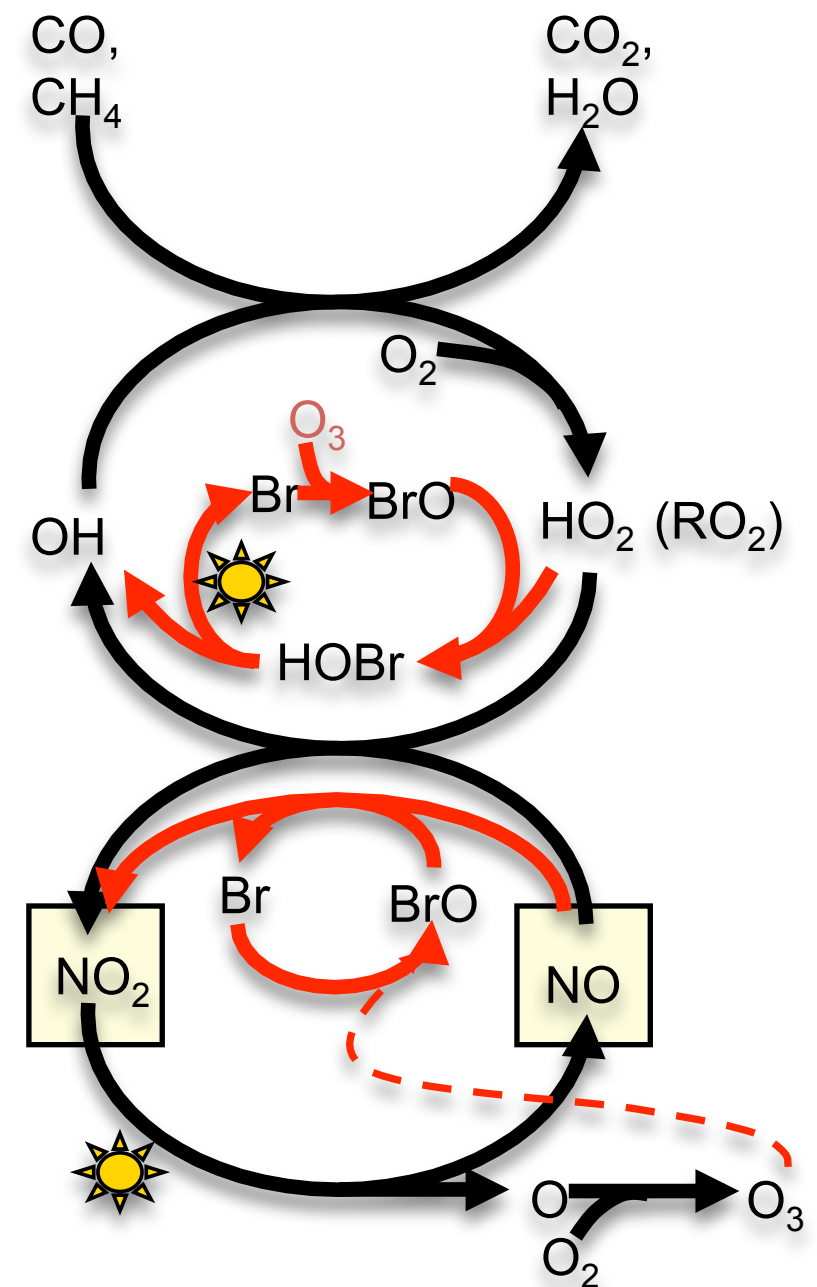
NO_x photochemistry over snow covered regions

- Nitrate photolysis in snow is a source of atmospheric NO_x
- NO_x plays a central role in atmospheric radical chemistry
- NO_x levels determine OH and O₃ concentrations, and thus the oxidative capacity of the atmosphere
- NO_x release from the snow pack could impact our understanding of the ice core record
- NO_x is closely coupled to other chemical cycles, i.e. halogen



Potential impact of a bromine source on the oxidation capacity of the atmosphere

- Reactive bromine impacts the most elemental atmospheric radical cycles
- Annually averaged 0.3 – 3 ppt BrO in the Arctic free troposphere (von Glasow et al., 2004) leads to:
 - ~10% reduction of free tropospheric O₃
 - 10-40% increase of the OH/HO₂ ratio and thus a change in methane lifetime
 - possible change in the Hg and DMS lifetime
- There is currently an incomplete understanding of atmospheric bromine sources



NO_x and bromine chemistry above snow have been investigated using field data from June 2008 and a 1D coupled snow atmosphere model

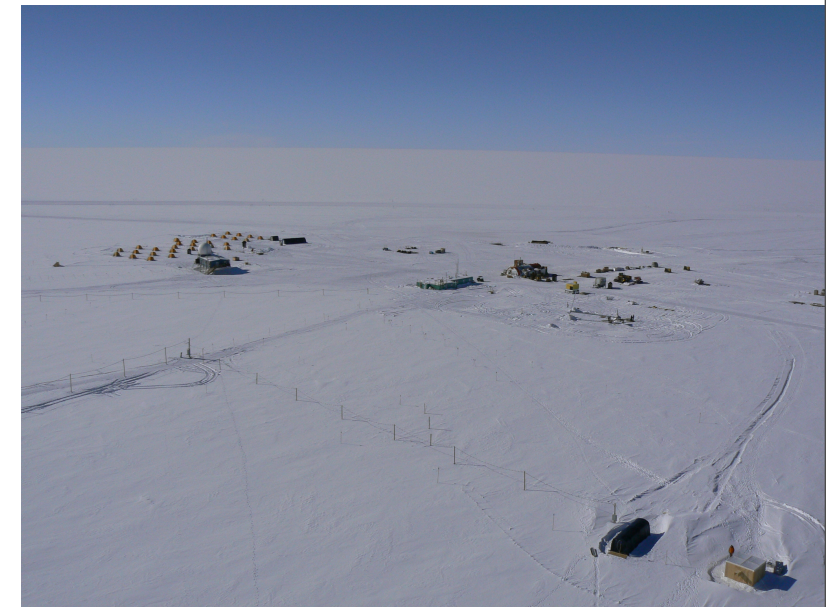
Greenland Summit Halogen and Hydrogen Radical Experiment (GSHOX)

- **Is there evidence for active bromine chemistry on the Greenland ice sheet?**
- **Does it influence atmospheric oxidative chemistry?**
- **Where does the bromine come from and where does it go?**
- **Can a 1D model more completely describe snow-atmosphere photochemistry?**

Field experiment :

- Spring 2007, Summer 2008 , at GeoSummit
- 6 research groups : UNH, UH, GaTech, UCLA, UCI, NOAA
- Comprehensive suite of measurements:
 - Hg (GEM, RGM, FPM)
 - OH, RO₂
 - NO, O₃
 - BrO
 - H₂SO₄, HCl, HO₂NO₂, SO₂, HCHO
 - Soluble Ions in air, firn, and surface snow (Br⁻, Cl⁻, etc.)
 - Aerosol size distribution
 - J-values air + firn
 - BL profiles (met)
 - HC, CO

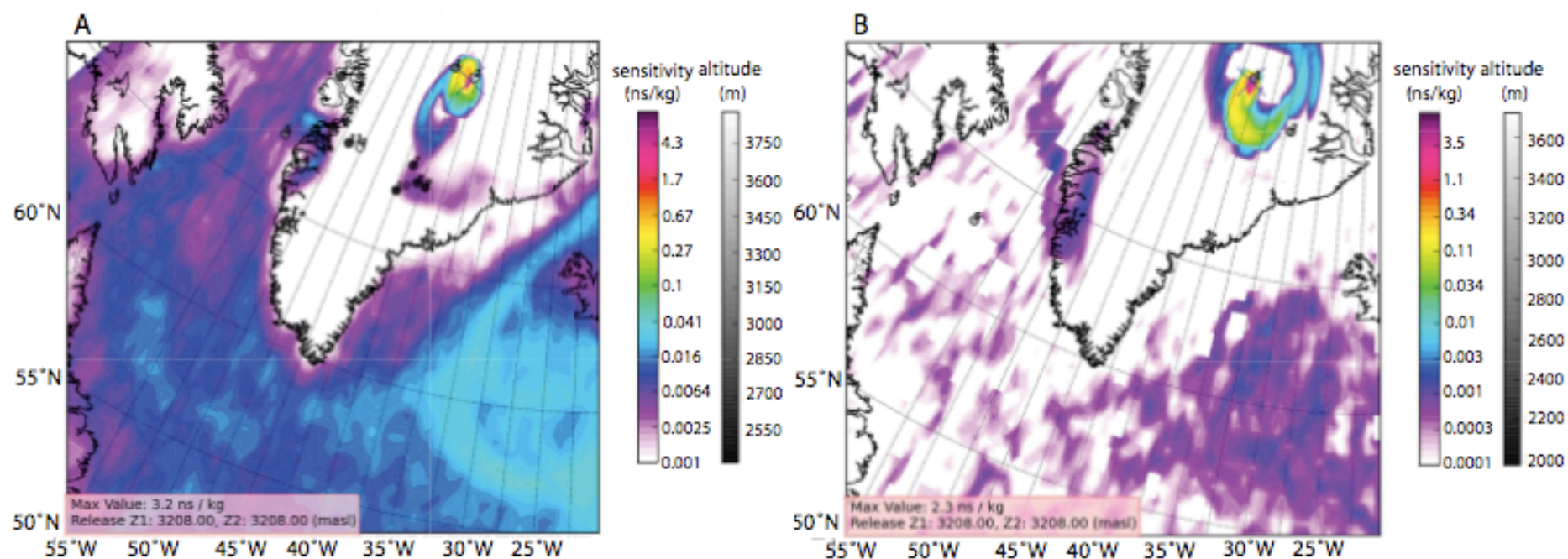
**See related poster #34 on
measurements during GSHOX!**



FLEXPART emission sensitivities indicate there are two types of air masses at Summit, Greenland

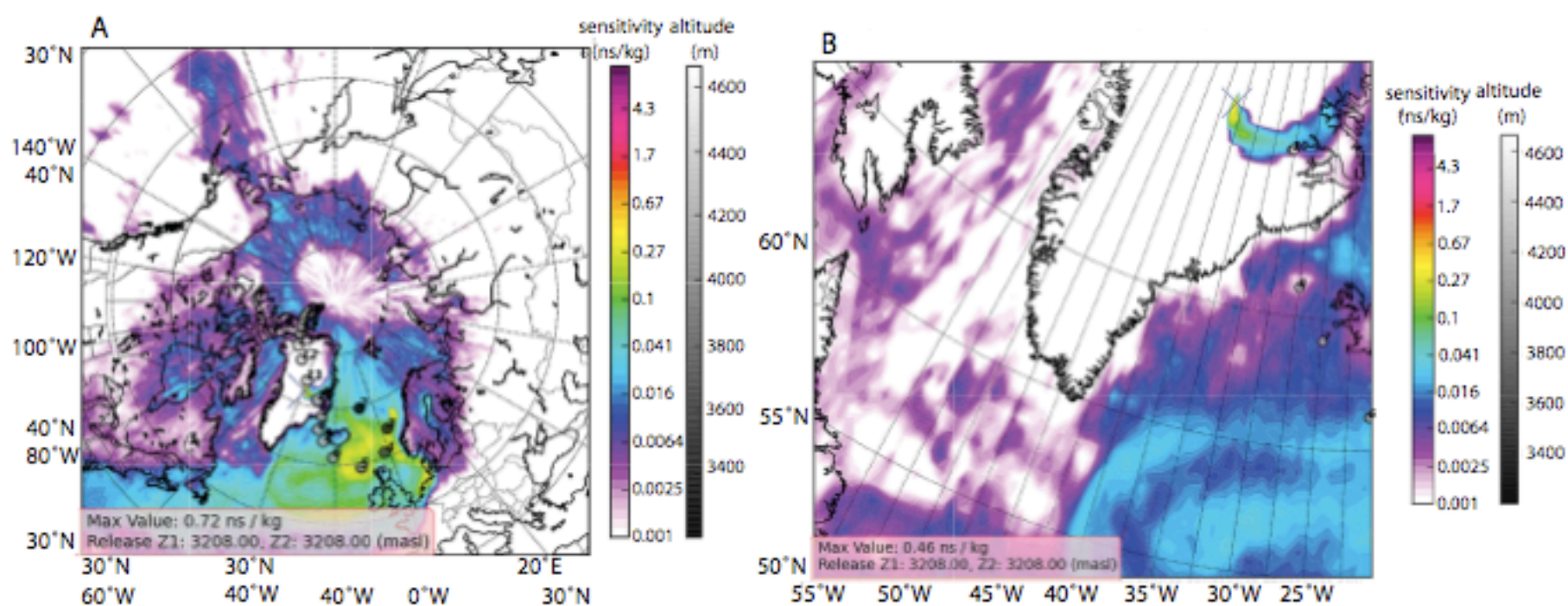
Local influenced

10 and 12 June 2008, 00:00–03:00UTC



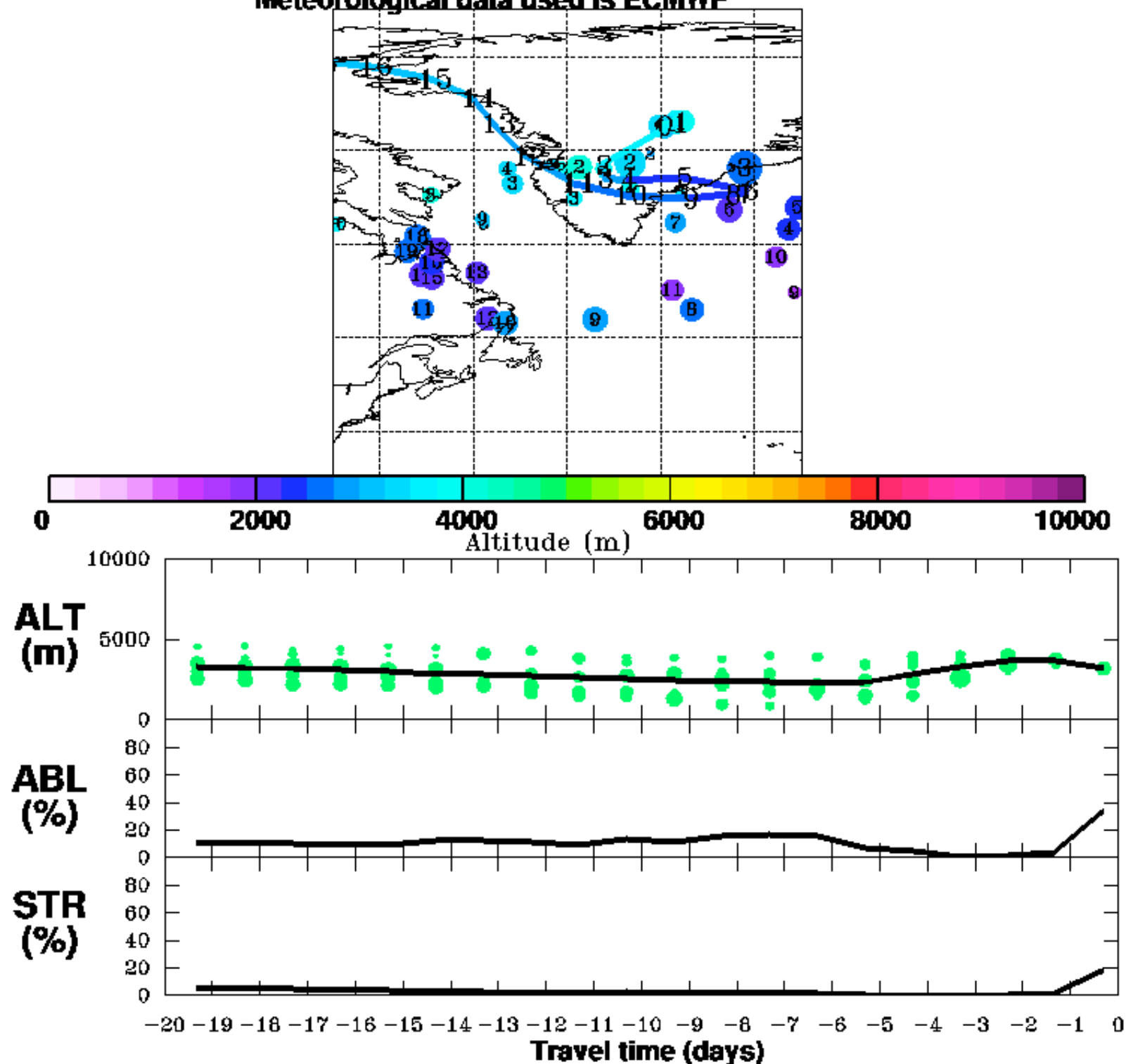
Marine influenced

4 July 2008, 12:00–15:00UTC



A UNIQUE FOCUS PERIOD FOR MODELING CAMPAIGN DATA: 10-13 JUNE 2008

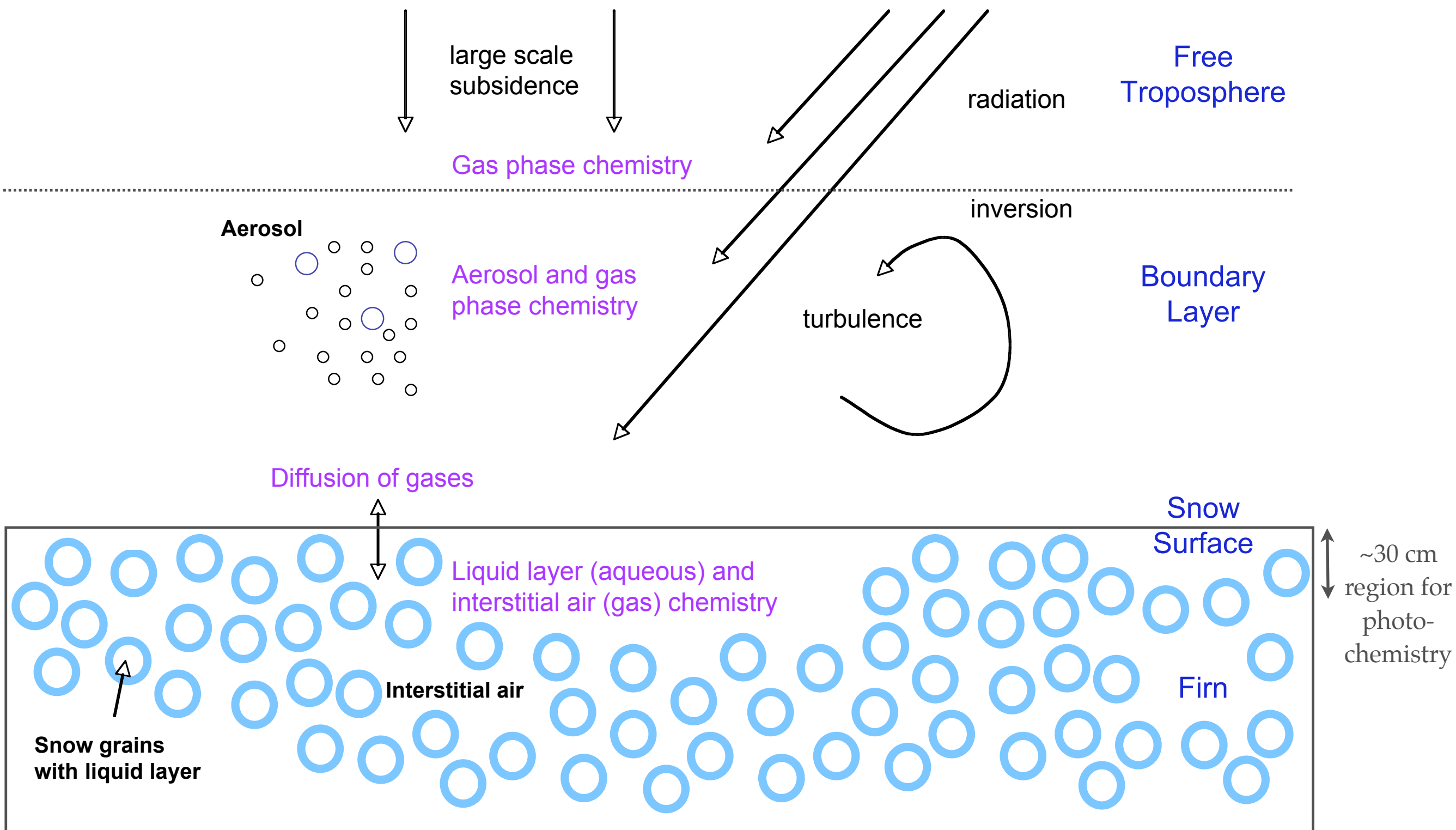
Retroplume summary for SUM_200806
Start time of sampling 20080610. 60000 End time of sampling 20080610. 90000
Meteorological data used is ECMWF



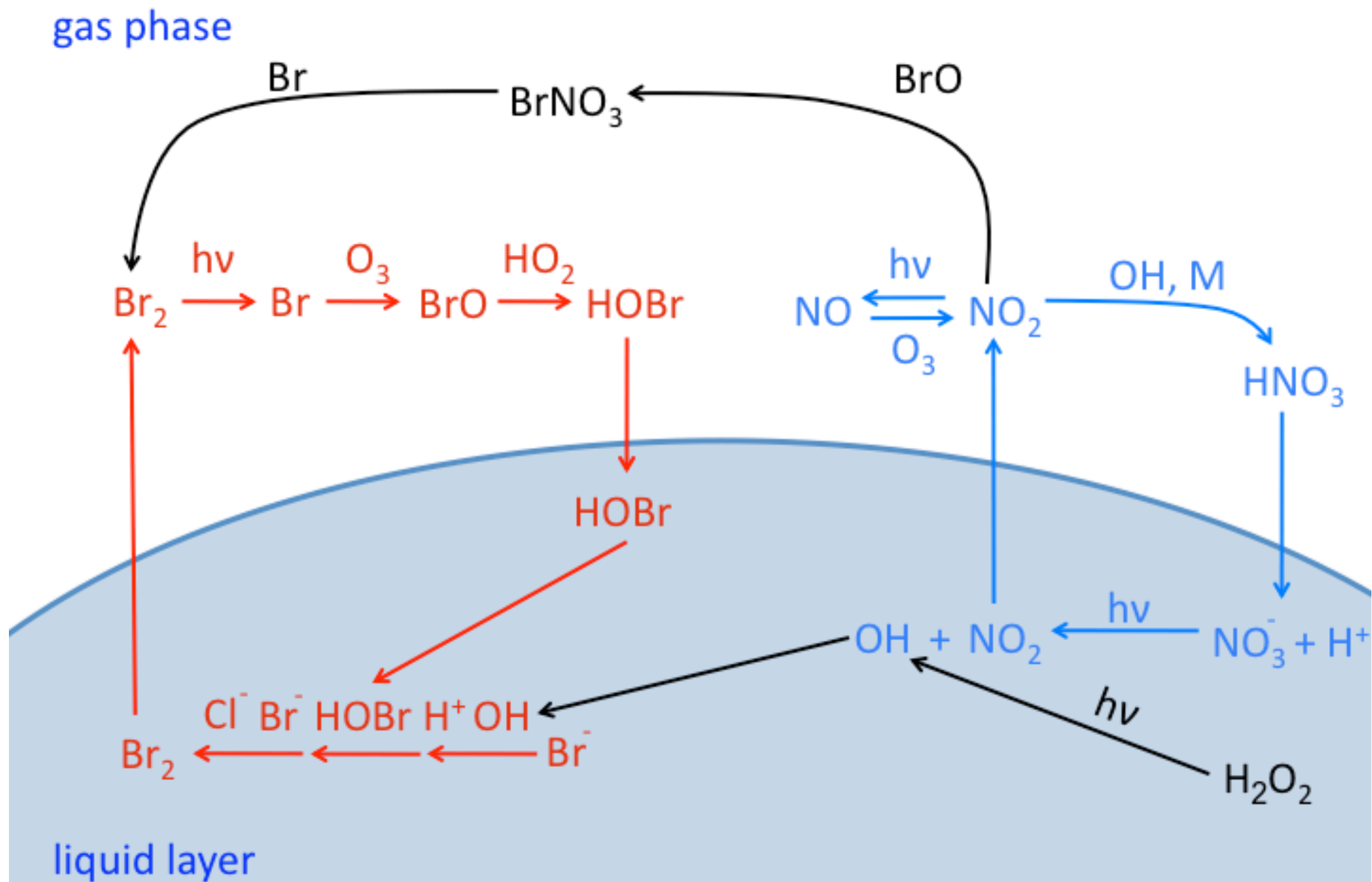
FLEXPART:
Retroplume
summary shows
the air resided
over Greenland
before reaching
Summit

Plot courtesy of John Burkhart

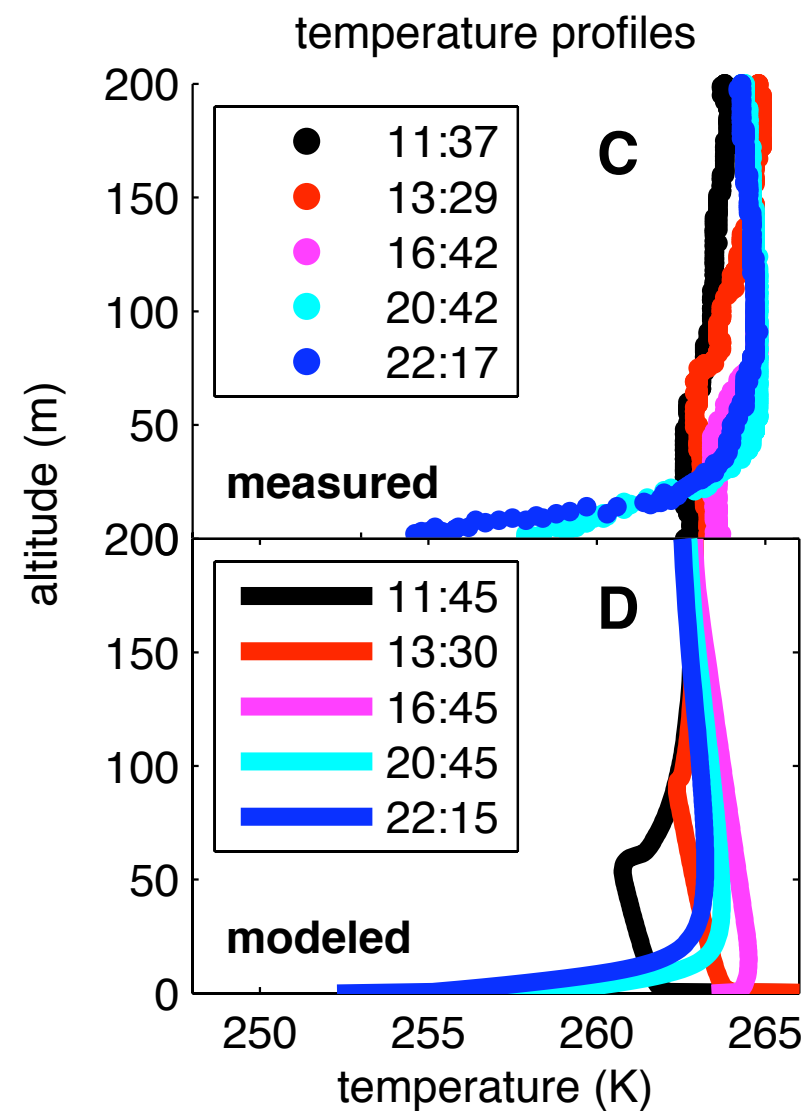
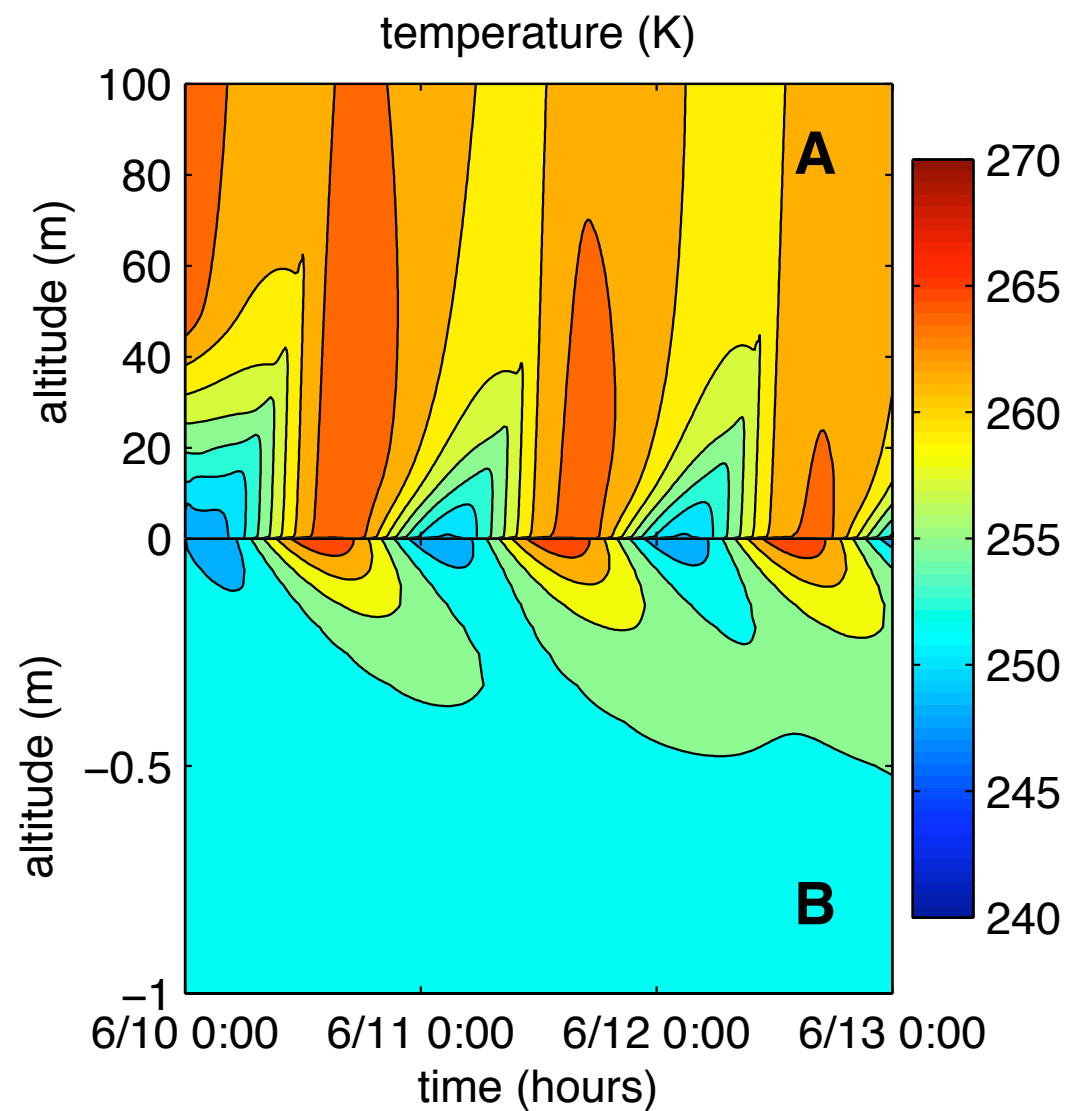
A SIMPLIFIED VIEW OF THE COUPLED SNOW-ATMOSPHERE SYSTEM



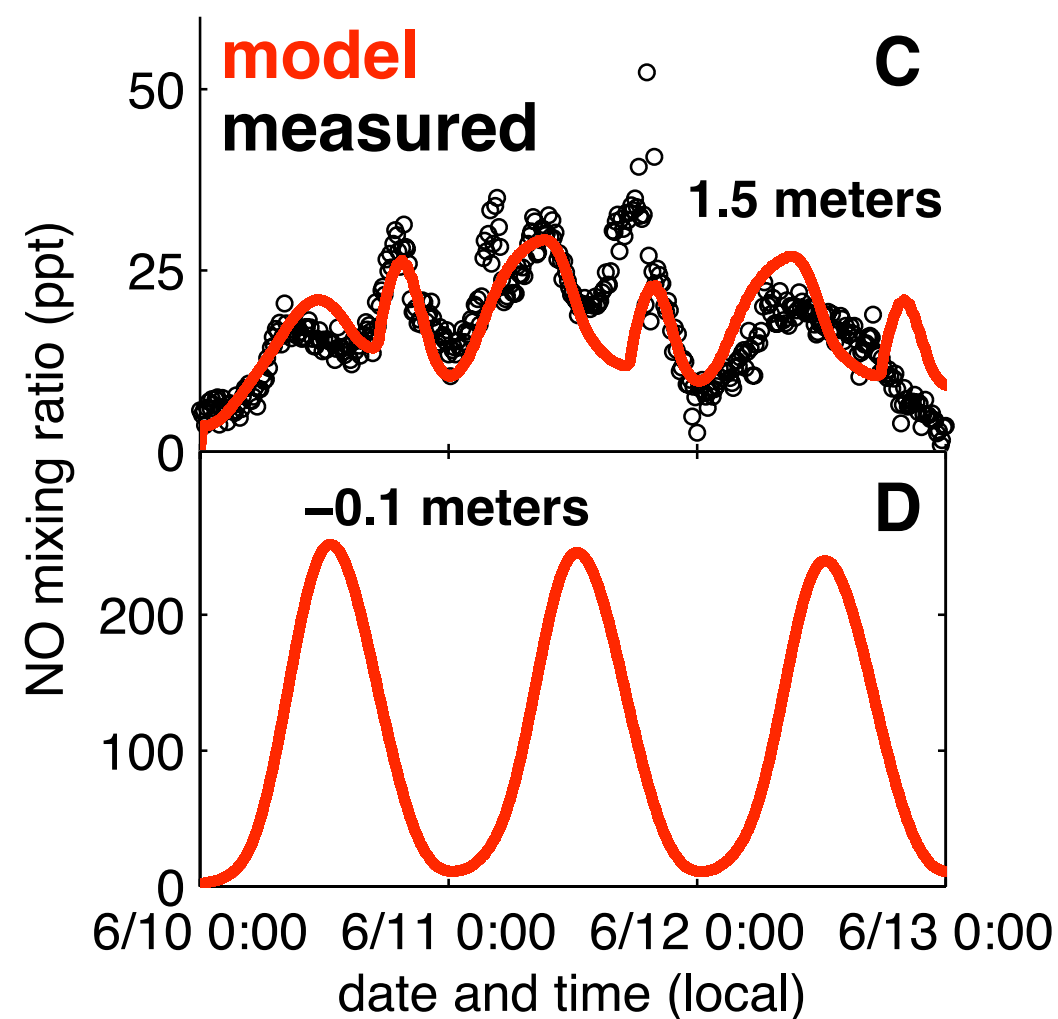
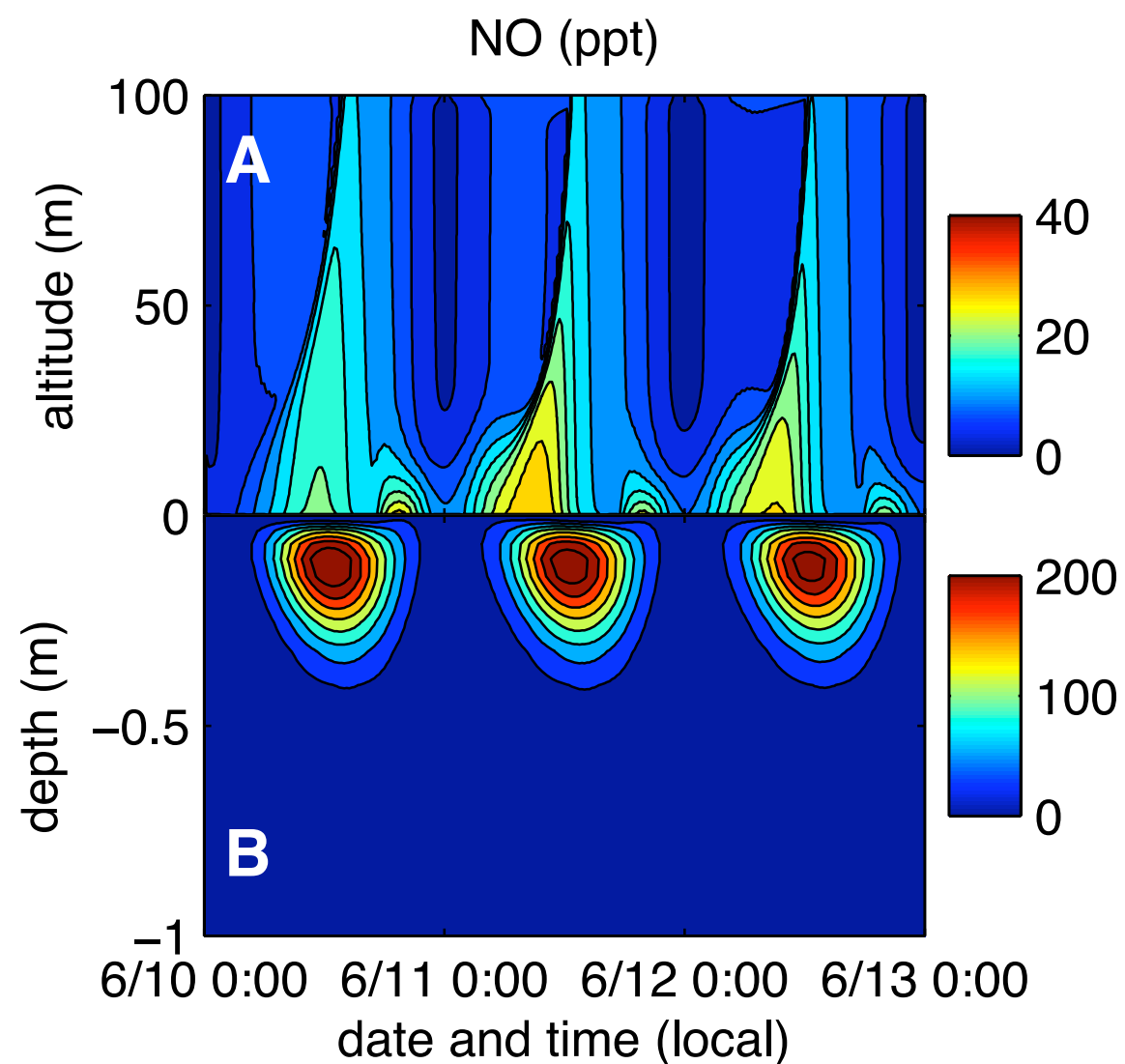
SUMMARY OF NO_x AND BROMINE CHEMISTRY OCCURRING AT SUMMIT



BOUNDARY LAYER EVOLUTION DURING THE FOCUS PERIOD IN 2008

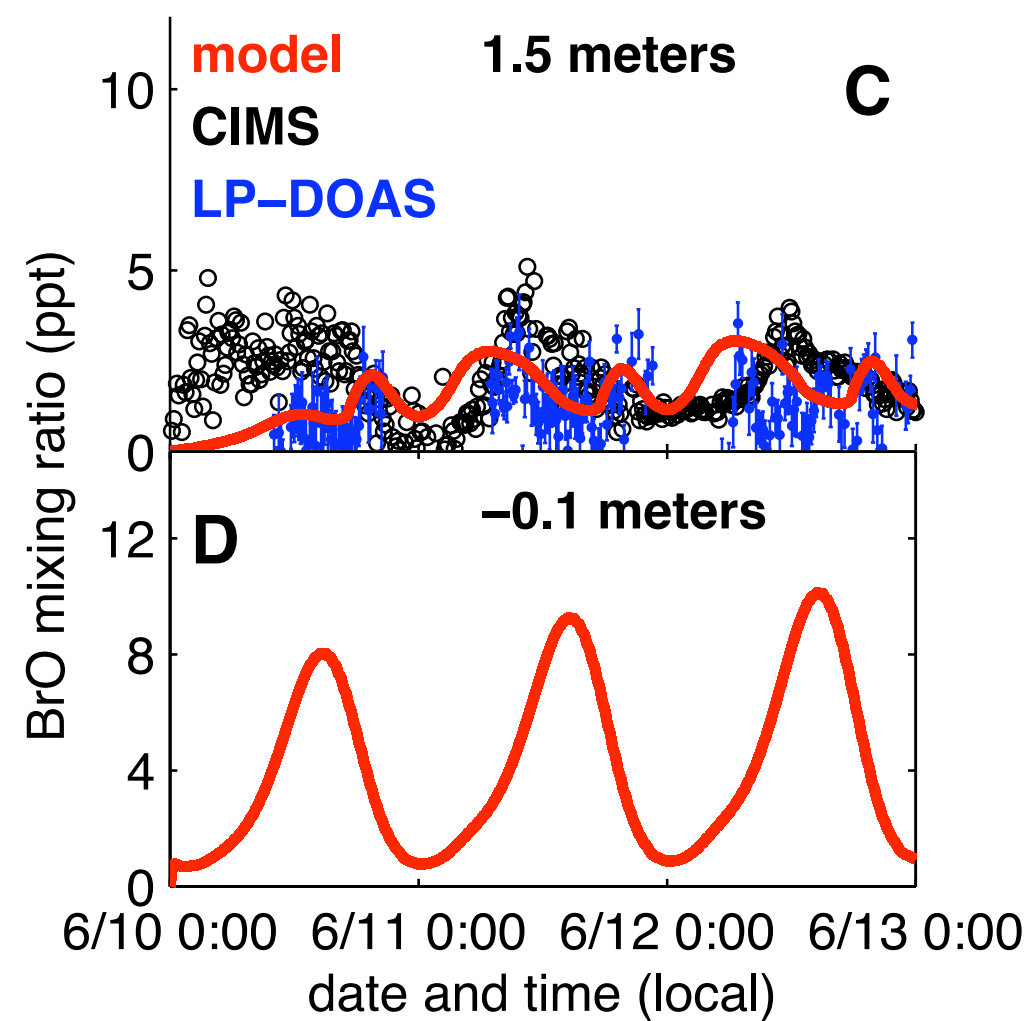
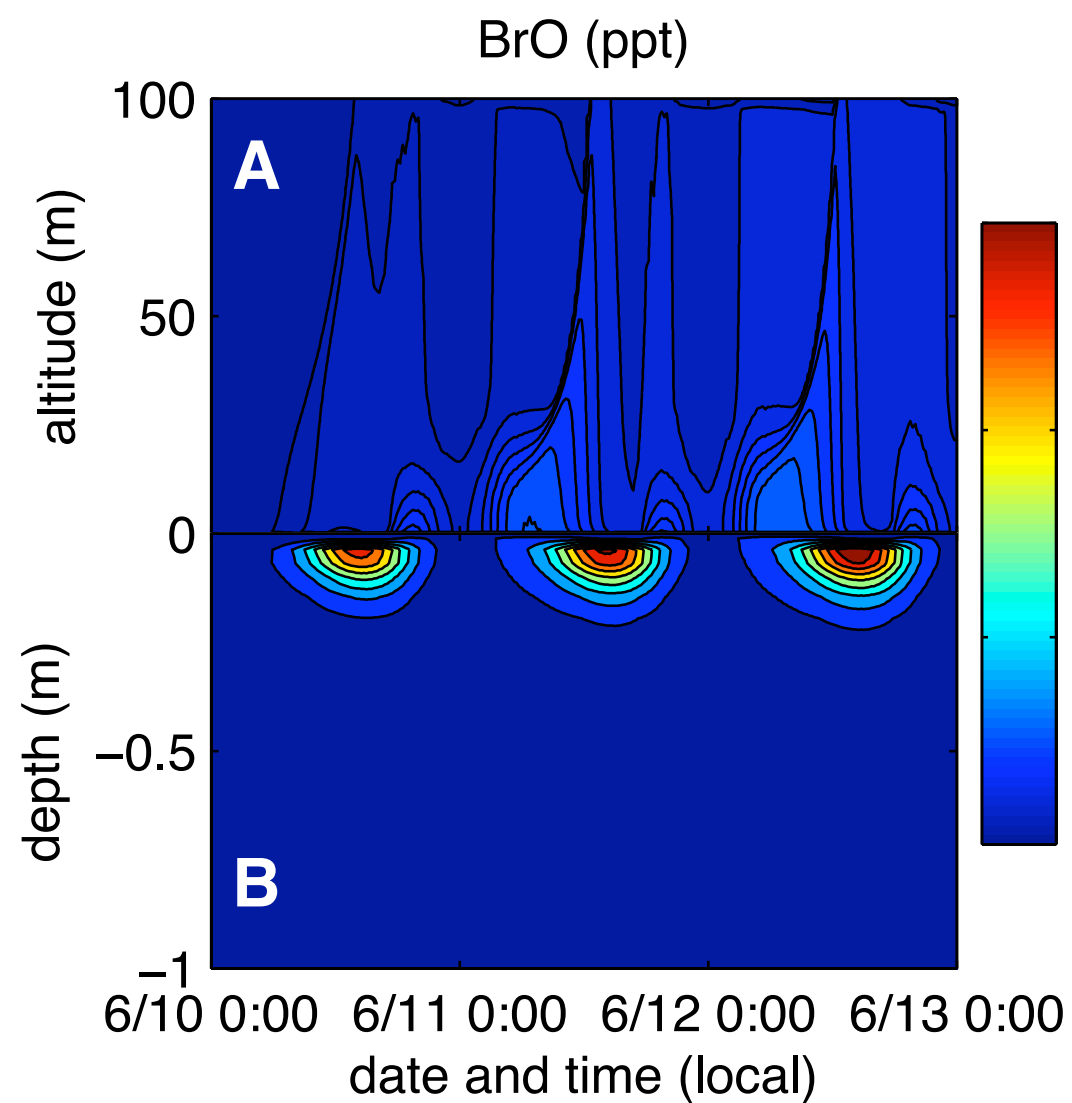


MODEL PREDICTED NO 10-13 JUNE 2008

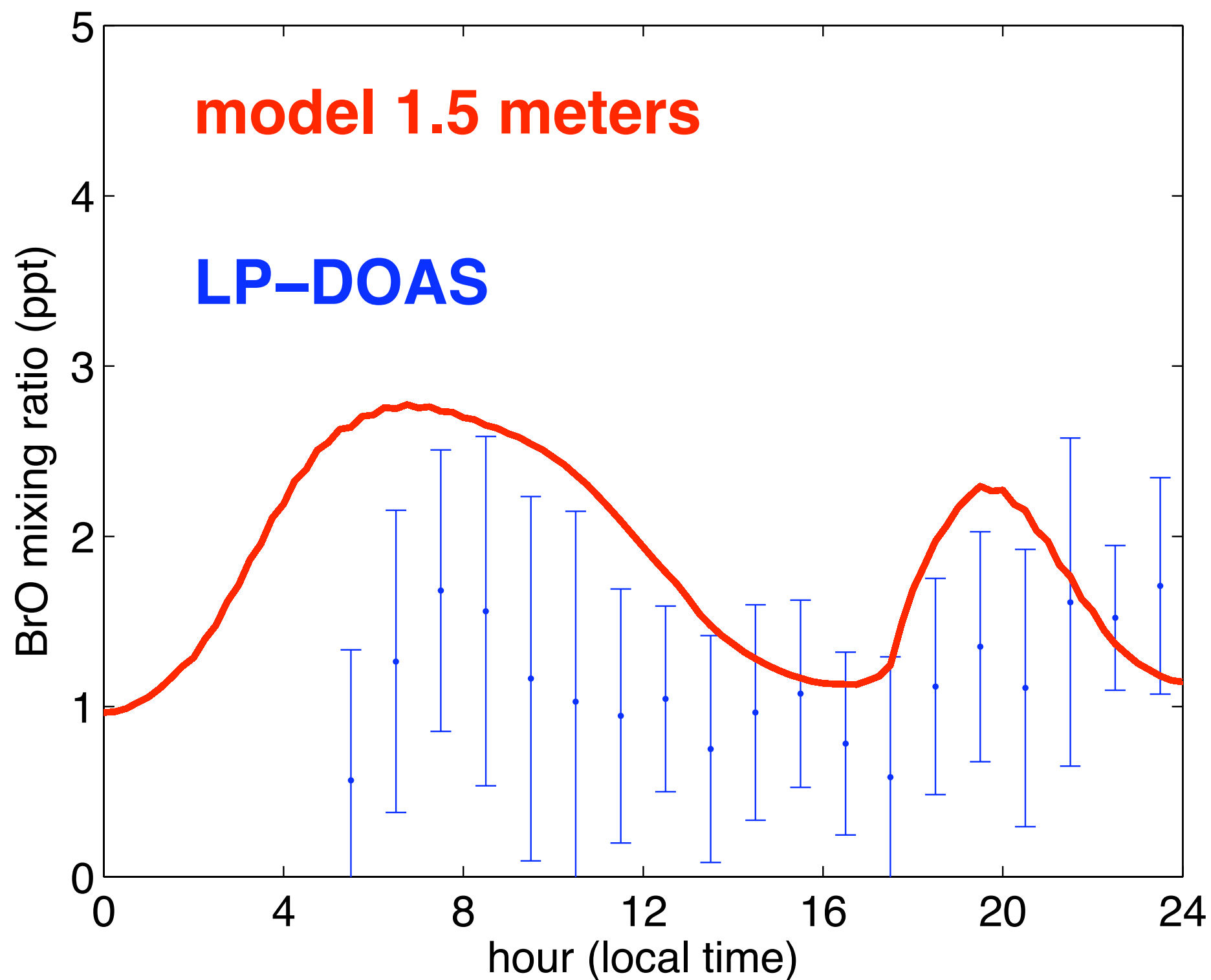


MODEL PREDICTED BRO

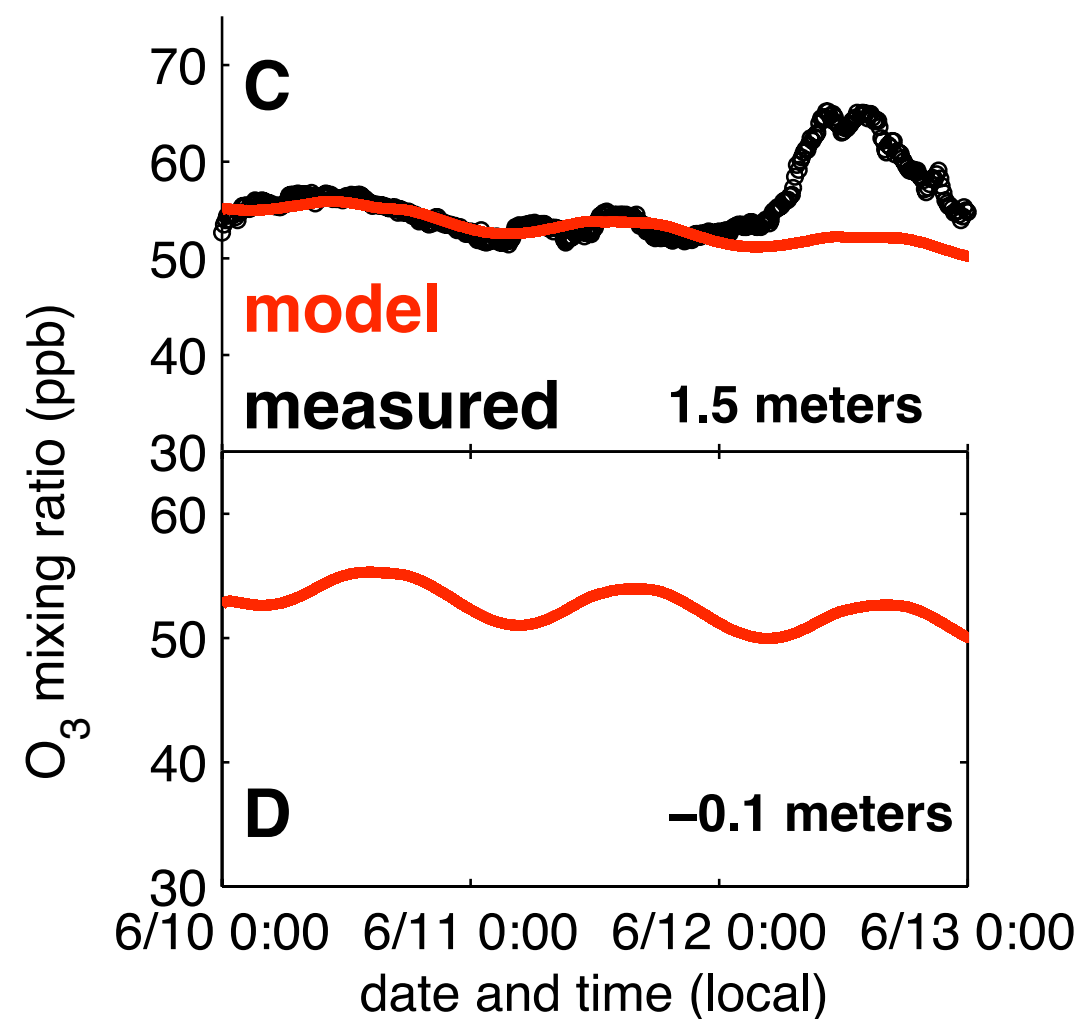
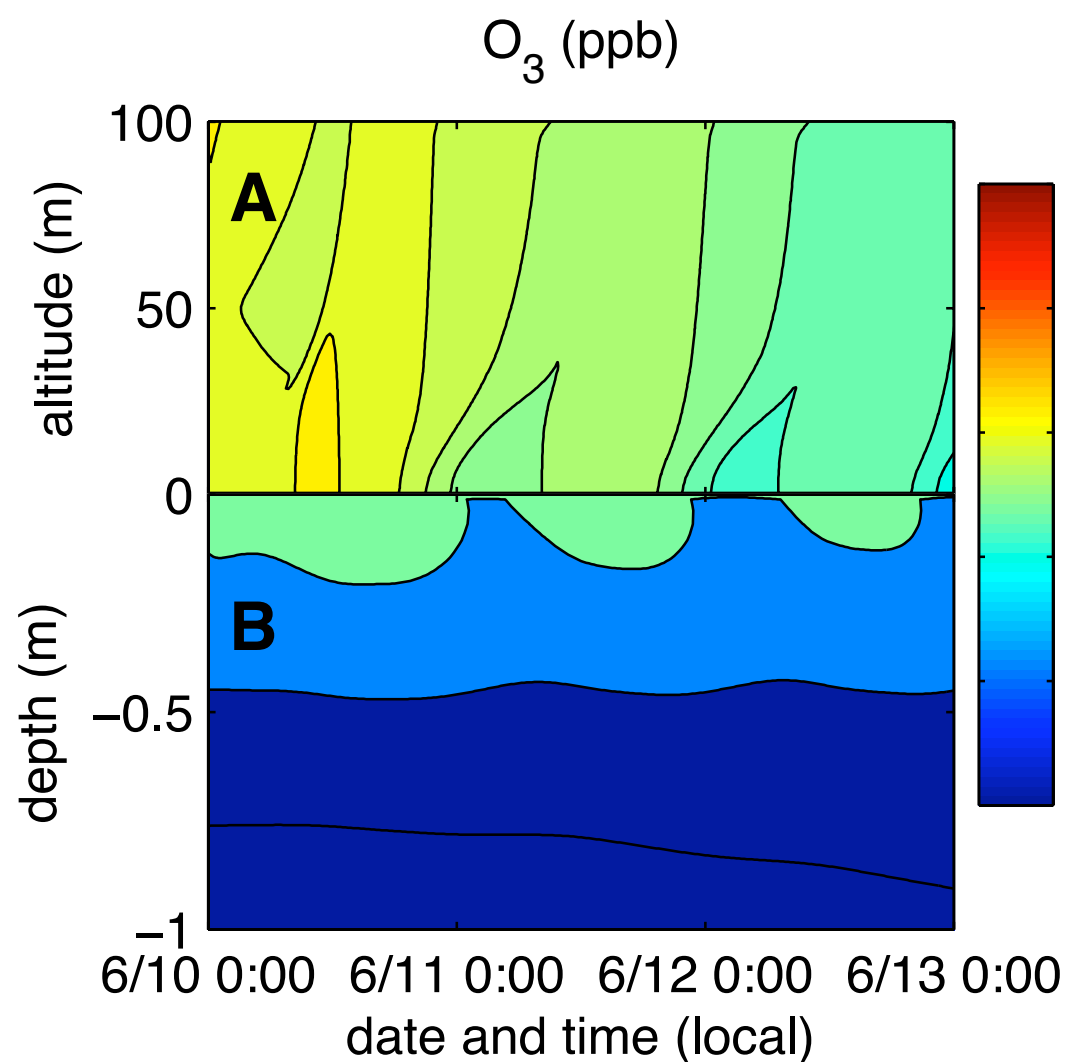
10-13 JUNE 2008



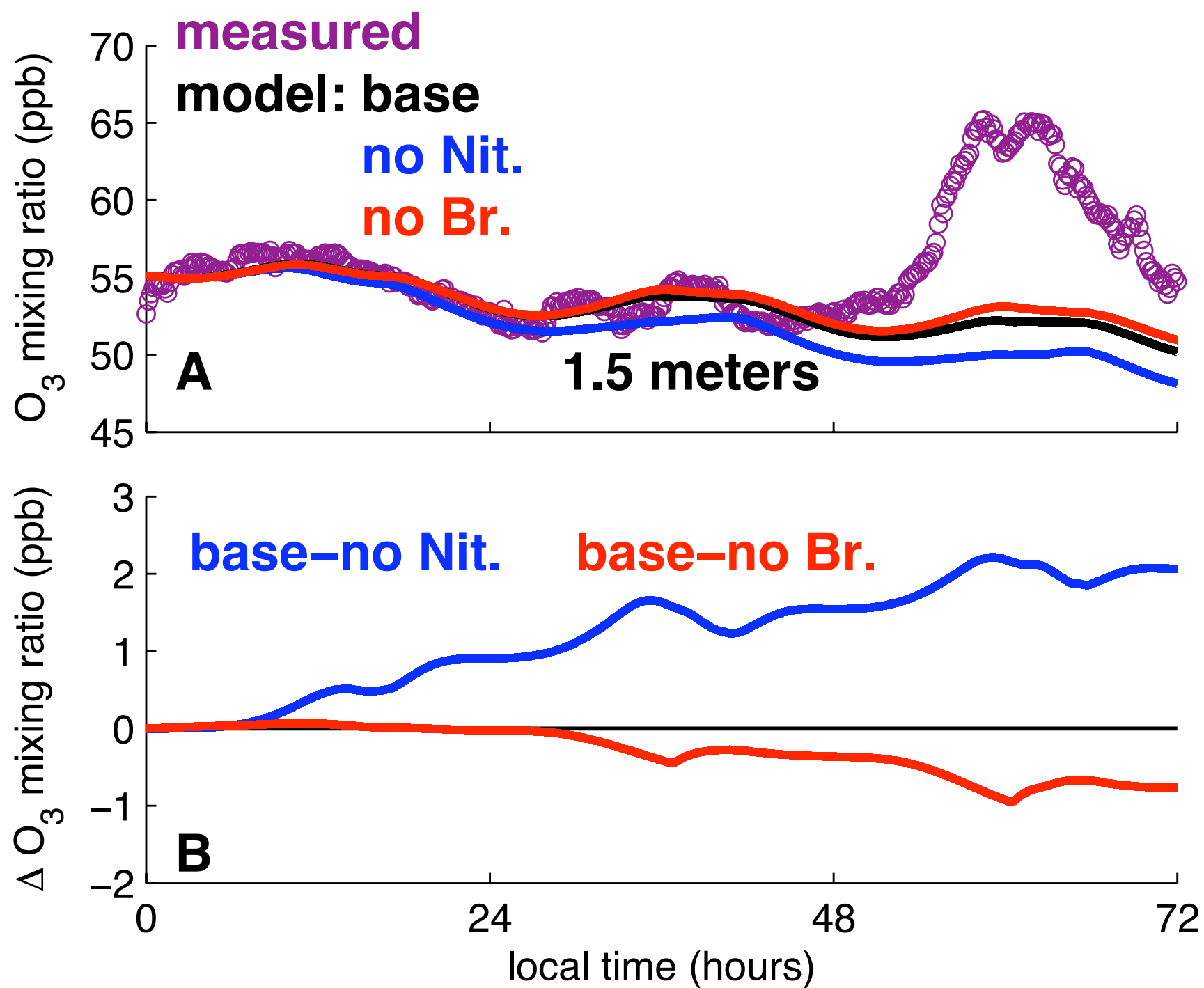
HOURLY AVERAGED BRO MODEL VS LP-DOAS MEASUREMENT



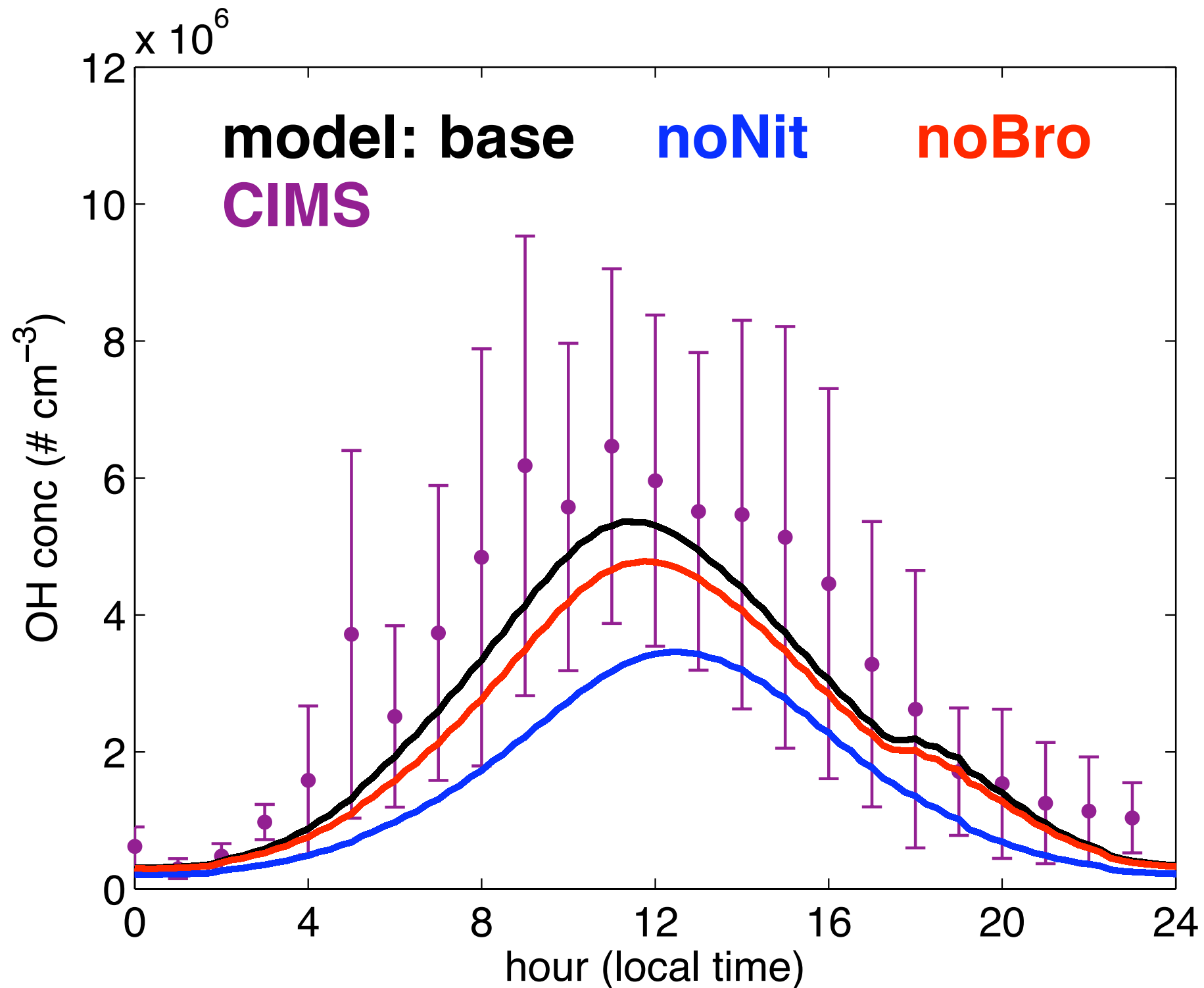
MODEL PREDICTED O₃ 10-13 JUNE 2008



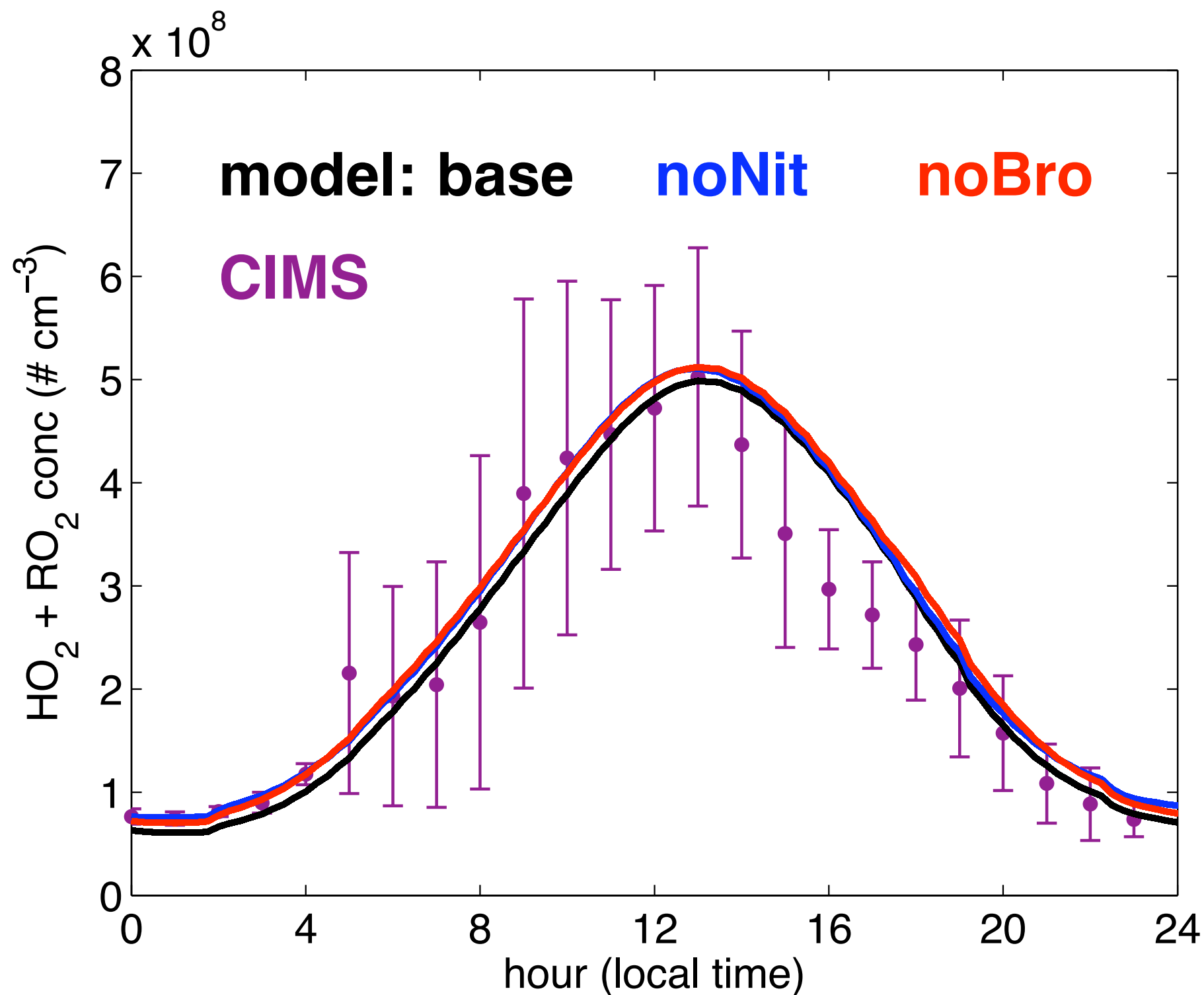
IMPACT OF SNOW DERIVED NO_x AND BROMINE CHEMISTRY ON OZONE



HOURLY AVERAGED OH MODEL VS CIMS MEASUREMENT



HOURLY AVERAGED RO₂ MODEL VS CIMS MEASUREMENT



CONCLUSIONS AND QUESTIONS FOR ONGOING WORK

- **Conclusions**

- NO_x and BrO mixing ratios in the BL at Summit can be described by a 1D model including snow chemistry.
- Snow derived bromine and NO_x impact both ozone and OH present in the boundary layer.

- **Questions**

- How does this focus period connect to long term trends at Summit?
- How are boundary layer processes impacting chemistry in the free troposphere?
- **Bromine**
 - Is gas phase bromine present in low concentrations at a wide range of non-coastal Arctic sites?
 - Does O₃ destruction by such small amounts of Br contribute the vertical profiles of O₃ at Summit (small depletion near the surface)?
- **Nitrogen**
 - Does nitrogen cycling in the BL impact nitrate measured in ice cores?
 - Does NO_x contribute to O₃ production, impacting the vertical profile of O₃ at Summit?

GSHOX SNOW PHOTOCHEMISTRY TEAM

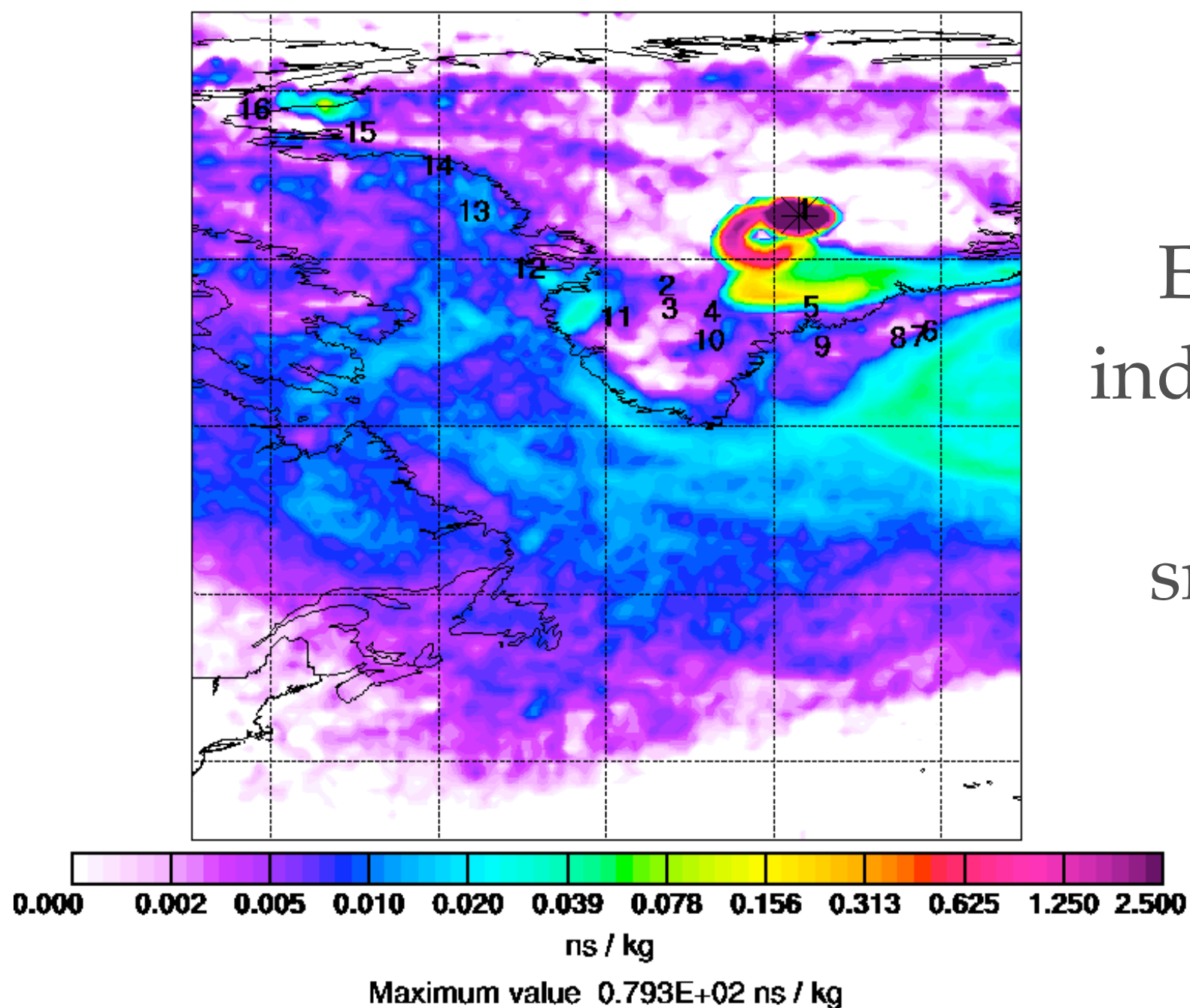
- To all the members of the GSHOX campaign



- To the members of the Stutz Group at UCLA.
- Detlev Helmig and Richard Honrath as well as their groups for collaboration and support.
- Zoe Courville for assistance.
- Kenjiro Toyota for assistance with implementing the wind pumping model.
- John Burkhart for the FLEXPART analysis.
- To the NSF GEO ATM Tropospheric Chemistry program for funding (Grant ATM-0612279:002). Logistic support was provided by NSF OPP Arctic Research Support and Logistics program and CH2MHill Polar Services.
- To the NY ANG for the heavy airlift, and permission to conduct research at Summit granted by the Danish Polar Center and Greenland Home Rule.

A UNIQUE FOCUS PERIOD FOR MODELING 2008 CAMPAIGN DATA: 10-13 JUNE

Footprint emission sensitivity in nested domain for SUM_200806
Start time of sampling 20080610. 60000 End time of sampling 20080610. 90000
Lower release height 3208 m Upper release height 3208 m
Meteorological data used are from ECMWF



FLEXPART:
Emission sensitivity
indicates measurements
are sensitive to
snowpack emissions